Exam. Code: 107402 Subject Code: 2114

B.Sc. (Bio-Technology) Semester—II BT-4: ORGANIC CHEMISTRY—B

Time Allc and -3 Hours]

[Maximum Marks-40

SECTION-A

Note: -- All questions are compulsory.

1. Complete the following reaction:

$$HC \equiv CH + H_2O \xrightarrow{i_2SO_4/HgSO_4}$$

- 2. How will you convert acetylene to 2-octyne?
- 3. Complete the following reaction:

$$H_3C - C - C - CH_3 \xrightarrow{CH_3MgBr} A \xrightarrow{H_2O} B$$

- 4. Explain Williamson's synthesis of ethers.
- 5. Why benzaldehyde is less reactive than acetaldehyde towards nucleophilic addition reactions?

6. Complete the following reaction and provide a suitable mechanism:

- 7. Acetyle chloride is more reactive towards nucleophilic subsubstitution reaction than ethyl chloride, why?
- 8. 2,4,6-trimethylbenzoic acid is difficult to esterify directly. Why? 1×8=8

SECTION-B

Note:—Attempt any FIVE questions. All questions carry equal marks.

- 9. Metal ammonia reduction of alkyne proceeds via entiaddition. Explain.
- 10. Alkynes are more reactive than alker es towards nucleophilic addition réactions. Explain.
- 11. Ethyl-tert-butyl ether can be prepared by reacting ethyl iodide with sodium tert-butoxide and not by reacting tert-butyl iodide with sodium ethoxide. Justify.
- 12. Discuss the mechanism of acid-catalyzed dehydration of alcohols to form ethers.
- 13. With mechanism, discuss how will you distinguish between 2-pentanone and 3-pentanone?

14. Provide suitable conditions and mechanism for the following conversion:

- 15. Write down the base-catalyzed mechanism of hydrolysis of esters.
- 16. Explain why decarboxylation in malonic acid is facile than acetic acid. 5×4=20

SECTION—C

Note: — Attempt any TWO questions. All questions carry equal marks.

17. (a) Complete the following reaction and provide a suitablemechanism:

$$H_2$$
 $H_3C - C = CT + H3r \xrightarrow{\text{Peroxide}} 3$

- (b) Discuss various factors responsible for the acidity of terminal alkynes.
- 18. (a) Complete the following reactions:

3

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3

(Contd.)

(b) Write a note on Crown Ethers.

3

19. (a) Provide a suitable mechanism for the following conversion:

(b) Complete the following reaction and provide a suitable mechanism

20. (a) Complete the following reaction and provide a suitable mechanism:

3

3

(b) Provide a suitable mechanism for the following conversion:

$$C_2H_5O$$
 OC_2H_5 C_2H_5ONa OC_2H_5 OC_2H_5 OC_2H_5ONa OC_2H_5 OC_2H_5

3

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